50 years of Neutron Backscattering Spectroscopy applied to molecular biology: 30 years "Protein Dynamical Transition" (1986)



W. Doster

Glass Transition of protein hydration water and structural flexibility of myoglobin in Structure, Dynamics and Function of Biomolecules (EBSA, Stockholm 1985) Springer Series in Biophysics Vol. 1 (1986) and

Wolfgang Doster, Anton Bachleitner, Rainer Dunau, Manfred Hiebl and Edgar Lüscher, Physics Department E 13, TUM Thermal Properties of Water in Myoglobin Crystals and Solutions at Subzero Temperatures, Biophysical Journal, 50, 213 (1986) "glass" transition of hydration water from the T-dependence of ir-OD, LMF and DR and specific heat around 200 K:





different onset temperatures interpreted by different instrument resolution

50 years of Neutron Backscattering Spectroscopy: A Protein Dynamical Transition (PDT) from Elastic Back-Scattering Displacements (1989) W. Doster



IN 13: D₂O-hydrated myoglobin, less water, PDT Fit



Water Coupled Protein Motions

why elastic scattering?

low cross section 200 – 600 mg sample

simple analysis: like Guinier SANS I_{el} = exp –(1/3Q² R²)

however: several transitions!

Is Protein Dynamics Relevant to Enzyme Catalyis? MD Simulations



Structural Fluctuations from Combination of QE Neutron Scattering and MD Simulation (since 1988)

Physica B 156 & 157 (1989) 437-443 North-Holland, Amsterdam CHAPTER II MACROMOLECULES AND BIOLOGICAL SYSTEMS INTERNAL DYNAMICS OF GLOBULAR PROTEINS: COMPARISON OF NEUTRON SCATTERING MEASUREMENTS AND THEORETICAL MODELS Jeremy SMITH', Krzysztof KUCZERA',*, Bruce TIDOR', Wolfgang DOSTER', Stephen CUSACK and Martin KARPLUS'

'Chemistry Department, Harvard University, 12 Oxford St., Cambridge, MA 02138, USA *Technische Universitiit Miinchen, Physik-Department E13, D-8046 Garching, Fed. Rep. Germany

3E. M. B. L., ClO1 ILL, Avenue des Martyrs, 156X, 38042 Grenoble Cedex, France invited paper







Dynamical transition of myoglobin revealed by inelastic neutron scattering

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four main novel features of bio-NS:

- study D_2O -hydrated proteins not solutions: reduce solvent scattering and protein diffusion, low T
- combination of BS and TOF,
- elastic and inelastic analysis: Dynamical Transition
- very wide temperature, Q and frequency range

Protein Dynamical Transition: Neutrons Scattering Nature (1989) with S. Cusack and W. Petry

UNCH



a second second



protein dynamics and function: 2 major components I, S

Doster et al. Nature 1989 Doster, Settles BBA 2005 Doster in Dynamics of Soft Matter 2011

1) internal: independent of solvent environment

non-Gaussian, discontinuous jumps, torsional jumps of side chains, methyl groups and main chain dihedral transitions

2) surface coupled motions: solvent viscosity Gaussian, water-assisted, small scale continuous translational displacements α- relaxation of hydration water

responsible for **PDT**









Heme displacements (Mössbauer)

Lichtenegger, Doster, Vogel et al. 1999 Biophys. J. 76,414 Effect of solvent: water, 75% glycerol/water, 80 % sucrose water



Heme PDT: onset temperature increases with solvent viscosity heme displacements are dominated by solvent motions near surface, less by internal protein dynamics!!

Onset temperature: solvent $\tau_c \cong 200 \text{ ns}$



Protein Function Kinetics of CO- binding to myoglobin different solvents



Log(CO unbound) (Kleinert, Doster et al. Biochem. (1998) 37,717) 0 В -1 S log N(t) -2 $\mathbf{T} = \mathbf{240} \ \mathbf{K}$ 60% E/W viscosity 75% G/W -3 90% G/W Δ glass 80% S/W dry -4 -8 -6 -4 -2 0 log (t / sec)

multiple binding steps: B,C (internal) S (from solvent)

Two kinds of elementary reaction rates

(Kleinert, Doster et al, Biochem. 1998), Doster Longeville in Dynamics of Soft Matter, Neutron Applications, Ch. 8 (2012) Springer

5

6



 $log\,(\,k_{BS}\,[s^{\text{-}1}]\,)$

What is a Protein Dynamical What is a Protein Dynamical Transition? GT and PT



NS: Incoherent intermediate scattering function I(Q,t,T, τ_c)



PDT (1)transition of the elastic intensity versus T at fixed resolution (2) plus: molecular process couples to solvent viscosity (3) no structural transition at T_D like the FST

"Dynamical" Transition in the Alanine Dipeptide in Crystal, Q, τ

an elastic transition occurs for each molecular process

Elastic Intensity (IN 13) and Model Fit with 3 methyls and dihedrals at Q = 4,3 A

Smith, Doster et al. JCP1996 Doster et al. JCP 2013





for each methyl group a step is observed in the elastic scattering, this is not a PDT because it is unrelated to solvent and viscosity



PDT α or β relaxation?

Mößbauer Effect : LMF, Fe in glycerol and myoglobin Doster et al. JCP 2013, Champeney, Woodhams, Nagel 1986

NBS: SPHERES, IN13: protein hydration water H₂O perdeuterated PC



0.1 ns

β = 1

0.1 ns

300

 T_{d}

250

 $\beta = 0.5$

350









PDT(Fe, H) reflects collective motions coupled to hydration water, observed at different time scales.

major heme motions are induced by solvent consistent with MD (Karplus et al.)

heme Mössbauer 200 K (140 ns) protein BS IN13 240 K (100 ps)

104(2010) 98101, JCP 139, 45105





Is Protein Dynamics Essential to Enzyme Catalysis?



enzyme catalysed reaction

reaction in solution

Arieh Warshel: Nobel Prize in Chemistry 2013, MD simulations of protein function

"The complex energy landscape is not the reason of the catalytic power of proteins! flexibility is interfering with rate acceleration, catalysis requires rigid stereo-chemical structures"

however:

this applies only to the catalytic step

the entrance and exit of ligands into enzymes require structural flexibility!!

Arieh Warshel and R.P. Bora Defining and quantifying the role of dynamics in enzyme catalysis J.Chem. Phys. 144, 180901 (2016)



- Two kinds of molecular processes in proteins identified:
 - 1) internal rotational transitions of side chains, CH₃ independent of protein environment and viscosity protein internal ligand displacements and reactions
 - 2) solvent viscosity dependent displacements coupled to translational diffusion of hydration water ligand exchange between solvent and active site
- PDT of heme iron and H < Δx^2 (T)> explained consistently: common cause: α -relaxation of solvent at different resolution
- no β relaxation in the hydration shell above 200 K
- no need to invoke a Frauenfelder energy landscape to explain the back-scattering spectra of protein dynamics and water!!

Collaborators



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